Er_2RhSi_3 and R_2CoGa_3 ($R \equiv Y$, Tb, Dy, Ho, Er, Tm, Yb) with Lu_2CoGa_3 type structure: new members of the AlB₂ structure family

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Abstract

The structure of Er₂RhSi₃ was redetermined by X-ray single-crystal diffraction (λ (Mo K α) = 0.71073 Å, μ =42.392 mm⁻¹, F(000) = 892, T = 293 K, R = 0.037, wR = 0.030 for 244 contributing unique reflections). It is shown that this silicide has a hexagonal structure of the Lu₂CoGa₃ type, hP24, (194) $P6_3/mmc$ -khfb, a = 8.1130(7), c = 7.7556(9) Å, V = 442.09(9) Å³, Z = 4, M_r = 521.68, D_x = 7.838 mg mm⁻³. The structure of isotypic Er₂Co_{1.4}Ga_{2.6} (hP24, (194) $P6_3/mmc$ -khfb, a = 8.607(3), c = 6.898(3) Å, V = 442.5(4) Å³, Z = 4, M_r = 598.30, D_x = 8.980 mg mm⁻³) was refined by X-ray powder diffraction (λ (Fe K α) = 1.93735 Å, F(000) = 1017.6, T = 293 K, R = 0.092 for 61 reflections). The cell parameters of R₂Co_{1+x}Ga_{3-x} phases (R = Y, Tb, Dy, Ho, Er, Tm, Yb; $x = \pm 0.4$) with the same structure type were obtained from X-ray powder diagrams. The Lu₂CoGa₃ structure is a distorted substitution variant of the AlB₂ type, where the trigonal Lu₆ prisms centred by cobalt atoms share triangular faces in infinite columns. The cobalt atoms are displaced away from the prism centres like the mercury atoms in the orthorhombic KHg₂ structure type. The distortions in Er₂RhSi₃ are of a lesser magnitude than those observed in Lu₂CoGa₃. A progressive substitution of gallium by cobalt along the RGa₂-RCoGa cross-sections of the R-Co-Ga systems (R = Y, Tb, Dy, Ho, Er, Tm, Yb, Lu) leads to a step by step deformation of the trigonal prisms. The main features of other deformation and substitution derivatives of the AlB₂ type are discussed.

1. Introduction

A series of rare-earth metal-rhodium silicides was reported by Chevalier *et al.* to crystallize with a new, AlB₂-related structure type [1]. The structure of Er_2RhSi_3 was refined from X-ray powder spectra and the authors state that the best agreement between calculated and observed line intensities was obtained for the space group $P\bar{6}2c$. However, a study of the atomic coordinates shows that the deviations from centrosymmetricity are very small indeed.

During systematic investigations of rare-earth metal-cobalt-gallium systems [2], the new compound Lu_2CoGa_3 was identified [3]. It crystallizes in the space group $P6_3/mmc$ with a structure which is very similar to that reported for Er_2RhSi_3 . We report here on the redetermination of the structure of Er_2RhSi_3 , motivated by its possible isotypism with Lu_2CoGa_3 . It also seemed of interest to extend the analysis to other members of the AlB₂ structure family and to study the effect of progressive cobalt substitution in some rare-earth metal digallides.

2. Experimental details

A sample of nominal composition Er₂RhSi₃ was prepared from high purity elements (erbium and rhodium 99.9%, silicon 99.999%) by arc melting under an argon atmosphere (the mass loss was 0.6%), annealing at 1073 K for two weeks in a silica tube under an argon atmosphere (5.3×10^4 Pa) and quenching in water. A single crystal (dimensions, \pm (100) 0.054 mm, \pm (010) $0.052 \text{ mm}, \pm (001) 0.026 \text{ mm}$) was mounted on a Philips PW1100 automatic four-circle diffractometer (Mo K α radiation, graphite monochromator). The unit-cell parameters refined from 2θ values of 21 reflections (Mo K α , $\lambda = 0.71073$ Å, $15^{\circ} < 2\theta < 35^{\circ}$) are in good agreement with those reported in ref. 1. 1042 reflections were collected out to $\sin\theta/\lambda = 0.702 \text{ Å}^{-1}$ ($0 \le h \le 9, 0 \le k \le 9$, $0 \le l \le 10$ and their antireflections) in the $\omega - 2\theta$ scan mode, yielding 275 unique reflections ($R_{int} = 0.036$). Two standard reflections, 202 and 220, showed a maximum intensity variation of 1.2%. An absorption correction was made using the program LSABS, described in ref. 4, with minimum and maximum transmission factors of 0.0757 and 0.1515. The anomalous dispersion coefficients were taken from ref. 5. Systematic absences led to the following possible space groups: $P6_3mc_1$, P62c and $P6_3/mmc$ [6]. The isotypism with Lu₂CoGa₃ was confirmed by the structure refinement, based on |F| values. 17 variables, including anisotropic atomic displacement parameters, were refined to R = 0.037 and

| Atom | Wyckoff position | x | у | Z | $U_{eq} 	imes 10^2$ (Å ²) |
|---|-----------------------------------|-------------------------|-----|-----------|---------------------------------------|
| $Er_{a}RhSi_{a}: a = 8.1130(7)$ | $r_{1}, c = 7.7556(9)$ Å, $V = 4$ | 42.09(9) Å ³ | | | |
| Si | 12(k) | 0.1669(2) | 2x | 0.0014(4) | 0.78(9) |
| Er(1) | 6(h) | 0.50914(5) | 2x | 1/4 | 0.75(3) |
| Rh | 40 | 1/3 | 2/3 | 0.0324(2) | 0.69(4) |
| Er(2) | 2(b) | 0 | 0 | 1/4 | 0.65(3) |
| $Er_2Co_{1.4}Ga_{2.6}$: $a = 8.60^{\circ}$ | 7(3), $c = 6.898(3)$ Å, $V =$ | 442.5(4) Å ³ | | | |
| Ga _{0.87(7)} Co _{0.13(7)} | 12(k) | 0.158(3) | 2x | 0.043(6) | 1.4(9) |
| Er(1) | 6(h) | 0.523(1) | 2x | 1/4 | 2.0(5) |
| Co | 4(f) | 1/3 | 2/3 | 0.065(6) | 1.3(8) |
| Er(2) | 2(b) | 0 | 0 | 1/4 | 2.2(6) |
| $Lu_2CoGa_3^{a}$: $a = 8.659(2)$ | 2), $c = 6.823(1)$ Å, $V = 44$ | 43.0(2) Å ³ | | | |
| Ga | 12(k) | 0.1692(2) | 2x | 0.0432(2) | 1.41(9) |
| Lu(1) | $6(\hat{h})$ | 0.5231(1) | 2x | 1/4 | 1.05(4) |
| Co | 4(f) | 1/3 | 2/3 | 0.050(1) | 1.6(1) |
| Lu(2) | 2(b) | 0 | 0 | 1/4 | 0.87(6) |

TABLE 1. Structure data for Er_2RhSi_3 , $Er_2Co_{1,4}Ga_{2,6}$ and Lu_2CoGa_3 (hP24, (194) P6₃/mmc-khfb, Z=4)

The equivalent isotropic atomic displacement parameters are expressed as $U_{eq} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$; for $Er_2 Co_{1.4} Ga_{2.6}$ isotropic displacement parameters were refined. *From ref. 3.

| Er ₂ RhSi ₃ | | $\mathrm{Er}_{2}\mathrm{Co}_{1.4}\mathrm{Ga}_{2.6}$ | | Lu ₂ CoGa ₃ | |
|-----------------------------------|-------|---|------|-----------------------------------|-------|
| Er(1)-2Rh | 2.992 | Er(1)-2Co | 3.05 | Lu(1)-2Co | 2.971 |
| 4Si | 2.999 | 4Ga* | 3.08 | 4Ga | 3.024 |
| 4Si | 3.084 | 2Co | 3.10 | 4Ga | 3.066 |
| 2Rh | 3.114 | 4Ga* | 3.15 | 2Co | 3.156 |
| 2Er(1) | 3.834 | 2Er(1) | 3.52 | 2Lu(1) | 3.481 |
| 2Er(1) | 3.886 | 2Er(1) | 3.71 | 2Lu(1) | 3.729 |
| 2Er(2) | 4.058 | 2Er(2) | 4.32 | 2Lu(2) | 4.343 |
| 2Er(1) | 4.279 | 2Er(1) | 4.90 | 2Lu(1) | 4.930 |
| Er(2)-6Si | 3.036 | Er(2)-6Ga* | 2.75 | Lu(2)–6Ga | 2.904 |
| 6Si | 3.050 | 6Ga* | 3.10 | 6Ga | 3.231 |
| 2Er(2) | 3.878 | 2Er(2) | 3.45 | 2Lu(2) | 3.412 |
| 6Er(1) | 4.058 | 6Er(1) | 4.32 | 6Lu(1) | 4.343 |
| Rh–3Si | 2.351 | Co-Co | 2.55 | Co-3Ga | 2.462 |
| 3Er(1) | 2.992 | 3Ga* | 2.62 | Со | 2.729 |
| 3Er(1) | 3.114 | 3Er(1) | 3.05 | 3Lu(1) | 2.971 |
| Rh | 3.375 | 3Er(1) | 3.10 | 3Lu(1) | 3.156 |
| Si-2Si | 2.345 | Ga*–2Ga* | 2.43 | Ga–Co | 2.462 |
| Rh | 2.351 | Co | 2.62 | 2Ga | 2.605 |
| 2Er(1) | 2.999 | Er(2) | 2.75 | Ga | 2.822 |
| Er(2) | 3.036 | Ga* | 2.86 | Lu(2) | 2.904 |
| Er(2) | 3.050 | 2Er(1) | 3.08 | 2Lu(1) | 3.024 |
| 2Er(1) | 3.084 | Er(2) | 3.10 | 2Lu(1) | 3.066 |
| Si | 3.856 | 2Er(1) | 3.15 | Lu(2) | 3.231 |

Ga* is Ga_{0.87}Co_{0.13}. E.s.d.s are less than 0.005, 0.07 and 0.010 Å for Er₂RhSi₃, Er₂Co_{1.4}Ga_{2.6} and Lu₂CoGa₃ respectively.

wR = 0.030 ($w = 1/\sigma^2(|F_{rel}|)$, S = 2.647), considering 244 contributing reflections with $|F_{rel}| > 3\sigma(|F_{rel}|)$. A secondary-extinction correction parameter (Gaussian distribution of mosaic spread) was refined to G = 0.0054(4). The maximum shift per e.s.d. in the last cycle was 0.2×10^{-4} and the final residual electron density +7.6(-7.1) electrons Å⁻³.* The programs used to

^{*}Further details of the structure determination (*e.g.* structure factors) can be ordered from the Fachinformationszentrum Karlsruhe, D-7514 Eggenstein-Leopoldshafen (deposition No 300248).

| R | a±0.003 (Å) | | c±0.003 (Å) | | $V \pm 0.4$ (Å ³) | |
|----|-------------|----------|-------------|----------|-------------------------------|----------|
| | x = -0.4 | x = +0.4 | x = -0.4 | x = +0.4 | x = -0.4 | x = +0.4 |
| Y | 8.727 | 8.660 | 7.169 | 7.014 | 472.8 | 455.5 |
| Tb | 8.764 | 8.680 | 7.095 | 7.037 | 471.9 | 459.2 |
| Dy | 8.740 | 8.653 | 7.048 | 6.994 | 466.3 | 453.5 |
| Ho | 8.735 | 8.630 | 7.012 | 6.945 | 463.3 | 447.9 |
| Er | 8.717 | 8.607 | 6.979 | 6.898 | 459.3 | 442.5 |
| Tm | 8.704 | 8.591 | 6.940 | 6.853 | 455.3 | 438.0 |
| Yb | | 8.858 | | 7.239 | | 491.9 |
| Lu | 8.652 | 8.560 | 6.801 | 6.785 | 440.9 | 430.6 |

TABLE 3. Unit-cell parameters for $R_2Co_{1+x}Ga_{3-x}$ phases (x = ±0.4) with Lu₂CoGa₃ structure type

refine the structure were all from the XTAL3.0 system [7]. The atomic positional parameters were standardized using the STRUCTURE TIDY program [8]. The final structural data for Er_2RhSi_3 are given in Table 1 and the interatomic distances in Table 2. Attempts to refine the structure in the non-isomorphic subgroup (190) $P\bar{6}2c$, chosen in ref. 1, did not improve the results and we conclude that the structure is centrosymmetric. The compounds with the same stoichiometry found in the {Y, La, Ce, Nd, Sm, Gd, Tb, Dy, Ho}–Rh–Si systems and reported to be isotypic with Er_2RhSi_3 [1] probably also crystallize in space group $P6_3/mmc$.

Rare-earth metal-cobalt-gallium compounds of compositions $R_2Co_{0.6}Ga_{3.4}$ and $R_2Co_{1.4}Ga_{2.6}$ (R = Y, Tb, Dy, Ho, Er, Tm, Yb) were also prepared by arc melting under an argon atmosphere. The samples were annealed at 873 K (ytterbium-containing samples at 673 K) for two weeks in a silica tube under vacuum and quenched in water. They were investigated by powder X-ray diffraction (diffractometer DRON-2.0, Fe K α radiation, $\lambda = 1.93735$ Å, internal silicon standard a = 5.4308 Å) and found to be isotypic with Lu₂CoGa₃. The unit-cell parameters are listed in Table 3. The structure of $Er_2Co_{14}Ga_{2.6}$ was refined from 61 reflections (40° < $2\theta < 108^{\circ}$) to R = 0.092 using programs from the PMNK system [9]. The standardized atomic parameters are given in Table 1 and the interatomic distances in Table 2.

3. Discussion

3.1. The crystal structures of the title compounds

Projections of Er_2RhSi_3 and Lu_2CoGa_3 along [0001] and [1120] are shown in Fig. 1. The Lu_2CoGa_3 type structure is a member of the AlB₂ family with an ordered arrangement of cobalt and gallium atoms inside face-sharing, distorted trigonal Lu_6 prisms. The ordering of the cobalt and gallium atoms is such that the cobaltcentred prisms share triangular faces to form infinite columns. The common faces are alternatively larger



Fig. 1. The structures of Er_2RhSi_3 , Lu_2CoGa_3 and YLiSn in projections along [0001] and [1120]. Large circles are erbium, lutetium or yttrium; medium circles are rhodium, cobalt or lithium; small circles are silicon, gallium or tin. Double full circles represent atoms of the same element which are superposed in projection in the translation unit.

and smaller and the cobalt atoms are displaced away from the centres of the prisms, along the 3-fold axes, towards the larger face. In this way, Co-Co dumb-bells appear with an interatomic distance of 2.73 Å. The gallium atoms are arranged in puckered hexagons (distance Ga-Ga 2.60 Å) which are loosely interconnected (distance Ga-Ga 2.82 Å) to form a tubular framework around the *c*-axis.

Isotypic compounds are found in the R-Co-Ga systems where $R \equiv Y$, Tb, Dy, Ho, Er, Tm, Yb [2]. Up to 13% of the gallium atoms in these R_2CoGa_3 compounds can be replaced by cobalt atoms. It can be seen from Table 1 that only the site 12(k) in $Er_2Co_{1.4}Ga_{2.6}$ has mixed Ga-Co occupation. On the other side of the ordered stoichiometry, substitution of up to 40% of the cobalt atoms by gallium atoms is possible and one expects to find a mixed occupation of the site 4(f). With increasing cobalt content, which has a smaller atomic radius than gallium, the unit-cell parameters decrease within the homogeneity range (see Table 3). In a similar way, when the atomic number of the rareearth metal increases (and the atomic radius becomes smaller owing to the lanthanide contraction), the unitcell volume decreases. It can be seen from Table 2 that the Co–Co dumb-bells in $\text{Er}_2\text{Co}_{1.4}\text{Ga}_{2.6}$ are even shorter (2.55 Å) than in Lu_2CoGa_3 , whereas the 3-fold connexion of the Ga* (*i.e.* Ga_{0.87}Co_{0.13}) atoms is less outstanding (distance inside hexagons 2.43 Å, distance between hexagons 2.86 Å) than in the fully ordered Lu_2CoGa_3 compound.

For Er_2RhSi_3 the distortions from the AlB₂-type atom arrangement with regular centred trigonal prisms are smaller than for the gallides. The rhodium atoms are only slightly displaced from the prism centres in such a way that the shortest Rh-Rh distance becomes 3.38 Å. The silicon atoms form isolated, approximately planar hexagons (distance Si-Si inside the hexagons 2.34 Å, shortest distance between hexagons 3.86 Å). An undistorted substitution variant of the AlB₂ type with the same atom ordering would crystallize in the space group P6/mmm-mfda, hP12, with cell parameters a' = a and c' = c/2 compared with Er_2RhSi_3 .

An ordered substitution variant of the Lu₂CoGa₃ type, reported for equiatomic YLiSn [10], is also presented in Fig. 1. The deformations are of similar magnitude to those observed for Lu₂CoGa₃ but the homogeneous ordering of the prism-centring lithium and tin atoms lowers the symmetry to space group $P6_3mc$. These atoms form a distorted tetrahedral wurt-zite-type framework, without homonuclear contacts.

The partly disordered structure of $CaZn_3$ [11] crystallizes with the same space group as Lu_2CoGa_3 and similar unit-cell parameters. It was described in terms of two types of atom arrangement, statistically distributed in the crystal. One of them is nearly identical with the structure of $BaLi_4$ [12], whereas the second, which was shown to be related to the KHg₂ type structure [13], is in fact a binary variant of the Lu₂CoGa₃ type.

3.2. Other deformation and substitution derivatives of the AlB_2 type

Figures 2, 3 and 4 show a series of other structure types belonging to the AlB_2 structure family. Those in Fig. 2 can be derived from the parent type by deformation, whereas Figs. 3 and 4 contain ordered substitution variants which are more or less strongly distorted with respect to the AlB_2 type. Some of the structures correspond to structure type branches which are generally separated in the literature [14, 15], *i.e.* AlB_2 and UHg_2 , EuGe₂ and CeCd₂, NdPtSb and LiGaGe, and HoNiGa, BiSCl and TiNiSi.

In the structure of AlB_2 [16], shown in Fig. 2, aluminium atom layers with triangle mesh (3⁶) alternate with boron atom layers with hexagon mesh (6³). The boron atoms are coordinated by tri-capped trigonal prisms of composition Al_6B_3 , whereas the aluminium atoms centre hexagonal B_{12} prisms, all faces of which are capped by aluminium atoms.



Fig. 2. The structures of AlB_2 and its binary deformation variants. Large circles correspond to the first element in the chemical formula.



Fig. 3. The structures of ternary substitution derivatives of the AlB_2 type (I). Large circles correspond to the first, medium circles to the second and small circles to the last element in the chemical formula.

Six different binary or pseudo-binary deformation variants of the AlB_2 type are known. For the UHg_2 [17] branch of AlB_2 the c/a ratio is significantly smaller than for AlB_2 itself (0.65 compared with 1.08). The



Fig. 4. The structures of ternary substitution derivatives of the AlB_2 type (II). Large circles correspond to the first, medium circles to the second and small circles to the last element in the chemical formula, except for BiSCl where the order is reversed. Asterisks indicate more than two superposed atoms in projection in the translation unit.

mercury atoms situated above and below the triangular faces of the prisms are at a contact distance to the central atom, the coordination number of which consequently increases to 11. In trigonal $CeCd_2$ [18] (c/a=0.68) and its branch EuGe₂ [19] (c/a = 1.22), the atoms centring the trigonal prisms (cadmium and germanium respectively) are displaced along the prism axis and form puckered hexagon mesh layers. The 3-fold connection of the germanium atoms in EuGe₂ is in agreement with the generalized 8-N rule for polyanionic valence compounds. In hexagonal CaIn₂ [20] similar puckered layers, formed by the indium atoms, are interconnected. Consecutive layers are related by a mirror plane and the *c*-parameter is doubled with respect to the cell of AlB₂. CaIn₂ is a Zintl compound where the indium atoms are arranged in a three-dimensional tetrahedral framework. In YAl1.4Si0.6 [21] and KHg2 [13] different distortions of the hexagon mesh layers, leading both structures to show orthorhombic symmetry, are observed. In KHg₂ the trigonal prisms are distorted in such a way that each prism has one smaller and one larger triangular face. The mercury atoms are displaced towards the larger face and form a threedimensional framework where each atom is surrounded by four others in a distorted tetrahedral coordination.

A number of ordered substitution variants of AlB_2 are known where the trigonal prisms are centred by two kinds of atom in the ratio 1:1. The simplest of these derivatives, which has the same number of atoms in the unit cell as AlB₂, was identified for SrPtSb [22] (c/a = 1.00). Two projections of this hexagonal structure, along [0001] and [1120], are shown in Fig. 3. Platinum atoms cap the rectangular faces of those trigonal Sr_6 prisms which are centred by antimony atoms and vice versa. The strontium atoms centre mixed hexagonal prisms of composition Pt₆Sb₆, capped by eight strontium atoms. In the orthorhombic distorted variant CaPtP [23], the deformations in the planes perpendicular to the prism axes are such that Pt-Pt pairs appear (2.85 A) and the hexagonal prisms surrounding half of the calcium atoms are transformed to pentagonal prisms.

In SrPtSb and CaPtP consecutive prism slabs are directly superposed. Four other types of stacking, with 2, 3, 4 and 6 prism slabs respectively, in the translation unit, have been observed up to now. In hexagonal ZrBeSi [24], presented in Fig. 3, consecutive layers formed by the prism-centring atoms (beryllium and silicon) are rotated by 60° around the *c*-axis with respect to each other. This stacking sequence can be denoted as BC. In the deformation derivative NdPtSb [25] the corresponding layers are slightly puckered, whereas for its branch LiGaGe [26] the distortions are so strong that a three-dimensional framework of gallium and germanium atoms is observed. Each gallium and germanium atom is surrounded by a Ge4 and Ga4 tetrahedron respectively, and the structure can be considered as a ternary substitution variant of CaIn₂. The YLiSn structure, shown in Fig. 1 and discussed above, is another deformation variant of ZrBeSi, where a distorted tetrahedral framework is also observed.

For the orthorhombic TiNiSi type [27] and its branches, as can be seen from their [010] projections given in Fig. 4, the atoms forming the trigonal prisms are displaced from their ideal positions. The HoNiGa [28] branch can be obtained by a ternary substitution from the binary KHg₂ type which implies a lowering of the symmetry from a body-centred to a primitive orthorhombic lattice. The nickel atoms in TiNiSi are inside almost regular Si₄ tetrahedra and, in comparison with HoNiGa, shifted so that infinite zigzag chains with an interatomic distance of 2.67 Å are observed. Of the six large atoms forming the distorted trigonal prism around the silicon site, one is at a considerably longer distance and does not belong to the first coordination sphere. The coordination of this site is generally described as a trigonal Ti₄Ni₂ prism capped by one titanium and two nickel atoms, the prism axis of this mixed prism being perpendicular to the axis of the Ti₆ prism. In a similar way, the coordination around the titanium site should be considered a penta-capped pentagonal prism of composition $Ti_4Ni_6Si_5$. The antitype BiSCl [29], a normal valence compound, is more closely related to the hexagonal Ni_2In type [30] and its orthorhombic distortion variant UPt₂ [31] where the majority atoms form the trigonal prisms corresponding to the aluminium atom prisms in AlB₂ and occupy half of the prism centres. The ternary isopointal substitution variant of UPt₂ is known as YAIGe [32].

A stacking with three trigonal prism slabs in the translation unit is found for trigonal CaLiSn [33] (see Fig. 4). Ignoring the displacements of the atoms away from the prism centres, similar to those observed in CaIn₂ and LiGaGe, the stacking sequence of the lithium and tin atom layers here is BCB. The lithium atoms and two thirds of the tin atoms are inside elongated tetrahedra of composition Sn₄ and Li₃Sn respectively, whereas the remaining tin atoms are surrounded by a trigonal Li₅ bipyramid. The four trigonal prism slabs in the translation unit of the hexagonal YPtAs type [25] are stacked in the sequence BBCC. The atom arrangement is close to that of AlB₂ and SrPtSb (c/4a = 0.89). The stacking in orthorhombic CaCuGe [34] corresponds to the sequence BBBCCC but, owing to the large distortions, the structure should rather be considered a substitution variant of KHg₂ and HoNiGa with a three times larger unit cell. No compound with undistorted slabs in the same stacking has yet been observed.

Another example of a substitution derivative of AlB_2 with non-equiatomic composition, $Tb_3Co_2Ge_4$ [35], is shown in Fig. 3. This monoclinic structure can be described as an intergrowth of TiNiSi- and AlB_2 -type columns parallel to [010]: $2TbCoGe + TbGe_2 =$ $Tb_3Co_2Ge_4$.

3.3. Effect of substitution on the crystal structure

Compounds crystallizing with one of the structure types discussed above contain 33.3 at.% of an element with a larger atomic radius. In the binary systems the compounds have the stoichiometry 1:2 and in the ternary systems they are to be found on the corresponding binary cross-section of the phase diagram. The majority of the rare-earth metal gallides and silicides RM₂ crystallize with structures of the AlB₂ family. Solid solutions are often observed when gallium or silicon is replaced by a small amount of a late transition metal (T). With a more extensive substitution, up to the composition RTM, new phases belonging to the same structure family may be obtained. Compounds with more than 33.3 at.% of transition metal generally adopt structures of the Laves-phase family, like the binary RT_2 compounds.

In each of the R–Co–Ga systems ($R \equiv Y$, Tb, Dy, Ho, Er, Tm, Yb, Lu) at least two compounds crystallizing with structure types of the AlB₂ family are found [2,

36, 37]. For example, in the Er-Co-Ga system studied at 873 K, four such compounds have been identified. Binary ErGa₂, with AlB₂-type structure (c/a = 0.96) [38], dissolves up to about 2 at.% cobalt. A second phase with an isopointal structure and a narrow homogeneity range is found at the composition ErCo_{0.2}Ga_{1.8}. For this compound the c/a ratio is considerably smaller (0.82) and corresponds to the UHg₂ type. Further substitution of gallium by cobalt leads to the formation of the Lu₂CoGa₃-type phase which has a homogeneity range from ErCo_{0.3}Ga_{1.7} to ErCo_{0.7}Ga_{1.3}. Inside the homogeneity range the c/a ratio is constant (0.80), but the unit-cell volume decreases with higher cobalt content (see Table 3). The equiatomic ErCoGa crystallizes with a HoNiGa-type structure [39]. The same structure types are observed in the {Y, Tb, Dy, Ho}-Co-Ga systems. The Tm-Co-Ga system is similar to the preceding systems but no compound occurs at the equiatomic composition and the gallide TmGa₂ crystallizes with a KHg₂ type structure [40]. On the corresponding crosssection of the Yb-Co-Ga system at 673 K only two compounds have been found: YbGa₂ with CaIn₂-type structure [41] and the ternary phase with Lu₂CoGa₃type structure.

In the same way as thulium, the binary gallide of lutetium forms a KHg_2 -type structure [40]. The hexagonal structure of Lu_3CoGa_5 ($LuCo_{0.3}Ga_{1.7}$) [42], which is an ordered substitution variant of the Fe₂P type [43], can also be derived from an AlB₂-type atom arrangement by considering a ternary substitution and a subsequent displacement of the cobalt atoms into the rare-earth metal atom layers. A narrow two-phase region separates this compound from the homogeneity range of the Lu_2CoGa_3 -type phase. At 873 K no compound is observed at the composition LuCoGa.

No systematic investigation of the Er-Rh-Si system has been carried out. A binary phase with AlB₂-type structure was only found for the off-stoichiometric composition $\text{ErSi}_{1.67}$ [44], whereas at high temperature ErSi_2 crystallizes with an α -ThSi₂-type structure [45], a stacking variant of AlB₂ with mutually perpendicular trigonal prism axes [46]. A HoNiGa-type structure was refined for the equiatomic compound ErRhSi [47]. No significant homogeneity range has been mentioned for Er_2RhSi_3 . The c/a ratio for this compound is 0.96, *i.e.* 11% lower than for the binary, silicon-deficient AlB₂type silicide (1.08).

In order to study the variation of the space-filling along the RT_2 -RTM cross-section in the Er-Co-Ga and Er-Rh-Si systems, the average volume per trigonal prism V_{6p} was calculated for each compound and plotted vs. the cobalt (rhodium) content (see Fig. 5). The volume of the distorted prisms in ErCoGa is 13% lower than that of the straight prisms in ErGa₂. From comparison of the atomic radii of cobalt and gallium it



Fig. 5. The average trigonal prism volume V_{6p} calculated for $\text{ErCo}_{y}\text{Ga}_{2-y}$ compounds, plotted as a function of the cobalt content $(0 \le y \le 1)$. The solid line corresponds to the homogeneity range of the Lu₂CoGa₃-type phase. Squares indicate the values calculated for $\text{Er}_{2}\text{RhSi}_{3}$ and ErRhSi_{3} .

can be estimated that a simple substitution of half of the gallium atoms by cobalt atoms inside straight prisms would lead to a smaller reduction in prism volume. The better space-filling is obtained by the step by step deformation of the ideal trigonal prisms. In the UHg₂type structure ErCo_{0.2}Ga_{1.8}, the trigonal prisms are compressed along the prism axis with contact distances Er-Er in this direction (3.58 Å). In the $Er_2Co_{0.6-1.4}$ -Ga_{3,4-2.6} phase three-quarters of the erbium atoms are displaced from their ideal positions and now have their contact distances in the (0001) plane (3.52 Å for $Er_2Co_{1,4}Ga_{2,6}$). For the remaining quarter of the erbium atoms the shortest distances are still observed along [0001] (3.45 Å). All erbium atoms are displaced in the ErCoGa structure and have short interatomic distances inside the distorted triangle mesh layers (3.36 Å) and between these layers (3.61 Å). A similar decrease in the average trigonal prism volume is observed between Er₂RhSi₃ with slightly distorted prisms and ErRhSi with strongly distorted prisms, even if in this case the atomic radius of the transition metal is slightly larger than that of the non-metal element.

It might be of interest to note that also for equiatomic CrB- and FeB-type compounds the volume of the trigonal rare-earth metal atom prisms decreases while main-group elements are replaced by transition metals at the prism centres [48].

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References

- 1 B. Chevalier, P. Lejay, J. Etourneau and P. Hagenmuller, Solid State Commun., 49 (1984) 753-760.
- 2 R. E. Gladyshevskii, *Thesis*, Moscow State University, Moscow, 1987.
- 3 R. E. Gladyshevskii and B. I. Chernyak, XI Ukrainian Republic Conf. on Inorganic Chemistry, Kiev, 1986, Collected Abstracts, p. 49.
- 4 E. Blanc, D. Schwarzenbach and H. D. Flack, J. Appl. Crystallogr., 24 (1991) 1035-1041.
- 5 J. A. Ibers and W. C. Hamilton (eds.), *International Tables* for X-ray Crystallography, Vol. IV, Kynoch, Birmingham, 1974 (present distributor Reidel, Dordrecht).
- 6 Th. Hahn (ed.), International Tables for Crystallography, Vol. A, Reidel, Dordrecht, 1983 (present distributor Kluwer, Dordrecht).
- 7 S. R. Hall and J. M. Stewart (eds.), XTAL3.0 User's Manual, Universities of Western Australia and Maryland, 1990.
- 8 L. M. Gelato and E. Parthé, J. Appl. Crystallogr., 20 (1987) 139-143.
- 9 V. K. Pecharskii, P. Yu. Zavalii, L. G. Aksel'rud, Yu. N. Grin' and E. I. Gladyshevskii, Vestn. L'vov. Univ., Ser. Khim., 25 (1984) 9-11.
- 10 G. Steinberg and H. U. Schuster, Z. Naturforsch., B34 (1979) 1165-1166.
- 11 M. L. Fornasini and F. Merlo, Acta Crystallogr., B36 (1980) 1739–1744.
- 12 F. E. Wang, F. A. Kanda, C. F. Miskell and A. J. King, Acta Crystallogr., 18 (1965) 24–31.
- 13 E. J. Duwell and N. C. Baenziger, Acta Crystallogr., 8 (1955) 705-710.
- 14 P. I. Kripyakevich, Structure Types of Intermetallic Compounds, Nauka, Moscow, 1977.
- 15 E. Hovestreydt, N. Engel, K. Klepp, B. Chabot and E. Parthé, J. Less-Common Met., 85 (1982) 247-274.
- 16 W. Hofmann and W. Jäniche, Naturwissenschaften, 23 (1935) 851.
- 17 R. E. Rundle and A. S. Wilson, Acta Crystallogr., 2 (1949) 148–150.
- 18 A. Iandelli and R. Ferro, Gazz. Chim. Ital., 84 (1954) 463-478.
- 19 E. I. Gladyshevskii, Dopov. Akad. Nauk Ukr. RSR, 2 (1964) 209-212.
- 20 A. Iandelli, Z. Anorg. Allg. Chem., 330 (1964) 221-232.
- 21 T. I. Yanson, Thesis, L'vov State University, L'vov, 1975.
- 22 G. Wenski and A. Mewis, Z. Anorg. Allg. Chem., 535 (1986) 110–122.
- 23 G. Wenski and A. Mewis, Z. Anorg. Allg. Chem., 543 (1986) 49-62.
- 24 J. W. Nielsen and N. C. Baenziger, U.S.A.E.C., (1953) 2728-2732.
- 25 G. Wenski and A. Mewis, Z. Kristallogr., 176 (1986) 125-134.
- 26 W. Bockelmann, H. Jacobs and H. U. Schuster, Z. Naturforsch., B25 (1970) 1305–1306.
- 27 C. Brink Shoemaker and D. P. Shoemaker, Acta Crystallogr., 18 (1965) 900–905.
- 28 Ya. P. Yarmolyuk, Yu. N. Grin' and E. I. Gladyshevskii, Dopov. Akad. Nauk Ukr. RSR, Ser. A, 9 (1979) 771–775.
- 29 E. Dönges, Z. Anorg. Chem., 263 (1950) 112-132.
- 30 E. S. Makarov, Izv. Akad. Nauk SSSR, Ser. A, 1 (1944) 29-33.

- 31 B. A. Hatt and G. I. Williams, Acta Crystallogr., 12 (1959) 655-657.
- 32 J. T. Zhao and E. Parthé, Acta Crystallogr., C46 (1990) 2276-2279.
- 33 W. Müller and R. Voltz, Z. Naturforsch., B29 (1974) 163-165.
- 34 W. Dörrscheidt, N. Niess and H. Schäfer, Z. Naturforsch., B32 (1977) 985–988.
- 35 P. K. Starodub, I. R. Mokraya, O. I. Bodak, V. K. Pecharskii and V. A. Bruskov, *Sov. Phys. Crystallogr.*, 31 (1986) 231–232.
- 36 Yu. N. Grin' and R. E. Gladyshevskii, *Gallides*, Metallurgia, Moscow, 1989.
- 37 R. E. Gladyshevskii, Yu. N. Grin' and Ya. P. Yarmolyuk, Stable and Metastable Phase Equilibrium in Metallic Systems, Nauka, Moscow, 1985, pp. 49-54.
- 38 S. E. Haszko, Trans. Metall. Soc. AIME, 221 (1961) 201-202.
- 39 O. M. Sichevich, R. E. Gladyshevskii, Yu. N. Grin' and Ya. P. Yarmolyuk, Dopov. Akad. Nauk Ukr. RSR, Ser. B, 11 (1982) 60-62.

- 40 E. I. Gladyshevskii, Yu. N. Grin', S. P. Yatsenko, Ya. P. Yarmolyuk and K. A. Chuntonov, Dopov. Akad. Nauk Ukr. RSR, Ser. A, 6 (1980) 82-85.
- 41 A. Iandelli, Z. Anorg. Allg. Chem., 330 (1964) 221-232.
- 42 R. E. Gladyshevskii, IV All-Union Conf. on Crystal Chemistry of Intermetallic Compounds, L'vov, October 18-20, 1983, Collected Abstracts, pp. 48-49.
- 43 S. Rundqvist and F. Jellinek, Acta Chem. Scand., 13 (1959) 425-432.
- 44 E. I. Gladyshevskii, Dopov. Akad. Nauk Ukr. RSR, 7 (1963) 886-888.
- 45 R. Nesper, H. G. von Schnering and J. Curda, VI Int. Conf. on Solid Compounds of Transition Elements, Stuttgart, June 12-16, 1979, pp. 150-152.
- 46 G. Brauer and A. Mitius, Z. Anorg. Chem., 249 (1942) 325-339.
- 47 W. Bazela, J. Leciejewicz and A. Szytula, Acta Phys. Pol., A68 (1985) 575-581.
- 48 E. Parthé, in M. O'Keeffe and A. Navrotsky (eds.), Structure and Bonding in Crystals, Vol. II, Academic Press, New York, 1981, Chapter 25, pp. 256–296.